

## Effect of long-term fertilizer application on yield and concentrations of elements (N, P, K, Ca, Mg, As, Cd, Cu, Cr, Fe, Mn, Ni, Pb, Zn) in grain of spring barley

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### ABSTRACT

Little attention has been paid to the effect of long-term fertilizer application on concentrations of elements in grain of barley produced on the metal non-contaminated agricultural soil. In 2010, we analyzed yield and concentrations of elements in grain of spring barley in unfertilized control, mineral fertilizer application ( $N_4P_2K_2$  – 70, 60 and 100 kg N, P and K per ha) and combinations of farmyard manure or poultry litter with mineral fertilizer ( $FMN_4P_2K_2$  and  $PLN_4P_2K_2$ ) treatments in the Ruzyně Fertilizer Experiment established on Luvisol in 1955 in Prague (Czech Republic). The yield of grain ranged from 4.03 to 9.74 t/ha in the control and  $FMN_4P_2K_2$  treatment. There was a positive effect of fertilizer application on concentrations of nitrogen, phosphorus and potassium, but no effect on concentrations of calcium and magnesium. With the exception of iron, concentrations of micro (copper and zinc) and risk elements (arsenic, cadmium, chromium, lead, manganese and nickel) were not significantly affected by the fertilizer treatments. Long-term use of organic and mineral fertilizers with appropriate application rates does not represent any risk for contamination of barley grain by risk elements on mineral rich and metal non-contaminated agricultural soils.

**Keywords:** arsenic; dilution effect; *Hordeum vulgare*; heavy metals; lead; nitrogen; zinc

The concentration of N in barley grain has largely been investigated (Grashoff and D'Antuono 1997, Sedlář et al. 2011), but little attention has been paid to concentrations of other elements, although they can also be important for evaluation of grain quality. For example an insufficient P concentration in seeds can negatively affect germination of seeds and growth of seedlings and therefore following yield of different crops (Hejzman et al. 2012a, Nadeem et al. 2012, White and Veneklaas 2012). High P concentration in seeds is thus highly important particularly for seeds for seeding.

Elements such as Ca, Cu, Fe, Mg and Zn are frequently lacking in human diet, therefore there is a high demand for their sufficiently high bio-available concentrations in cereals (White and

Broadley 2009). Decrease in concentrations of Cu, Fe and Zn in the cereal grain occurred during the last 50 years due to introduction of short straw and high grain yielding varieties. Highly topical question is therefore biofortification – methods enabling to increase bio-available concentrations of Ca, Cu, Fe and Zn in edible crops.

The adverse effects of many risk-elements such as As, Cd and Pb on human health have led to evaluation of agricultural products not only according to basic quality parameters, but also according to concentrations of risk-elements (Dudka et al. 1996, Šrek et al. 2012).

Concentrations of macro-, micro- and risk-elements in the barley grain have been frequently investigated in highly polluted regions but little

attention has been paid to the effect of long-term fertilizer application on common agricultural soils. In addition, little attention has been paid to relationship between grain yield and concentration of elements. The aim of this study was therefore to investigate how the grain yield of spring barley and concentrations of elements (N, P, K, Ca, Mg, As, Cd, Cu, Cr, Fe, Mn, Ni, Pb, Zn) in the grain are affected by long-term fertilizer application.

## MATERIAL AND METHODS

**Experimental design.** The Ruzyně Fertilizer Experiment (RFE) was established on a permanent arable field in 1955 on the western edge of Prague (50°05'15"N, 14°17'28"E). At the study site, the mean annual temperature was 8.2°C and the mean annual precipitation was 422 mm (Prague-Ruzyně Meteorological Station, 1955–2007). The soil type was classified as Luvisol and the parent material was loess mixed with chalk. The upper 30 cm (arable layer) contains 27% clay, increasing to 40% in the subsoil (soil layer 30–40 cm) and 49% at depths of 40–50 cm (Hejčman et al. 2012b).

Spring barley was only sampled from strip number II of the RFE, named the 'Classical Crop Rotation' (45% cereals, 33% root crops and 22% legumes). The crop sequence was alfalfa, alfalfa, winter wheat, sugar beet, spring barley, potatoes, winter wheat, sugar beet and spring barley with alfalfa under-sowing. Organic fertilizers, farmyard manure (FM) and poultry litter (PL), were applied three times during the nine years long crop rotation in each autumn before planting of sugar beet or potatoes in particular treatments. The last application of FM and PL was performed before sugar beet planting one and half year before the seeding of spring barley investigated in this study

(FM and PL applied in October 2008). Calcium ammonium nitrate (27% N), superphosphate (8.3% P) and potassium chloride (49.8% K) were applied as mineral fertilizers.

We selected unfertilized control, mineral N, P and K fertilizer application ( $N_4P_2K_2$ ) and combinations of organic fertilizers with mineral N, P and K fertilizers ( $FMN_4P_2K_2$  and  $PLN_4P_2K_2$ ). Each treatment was replicated four times (in detail Šrek et al. 2010). The individual plot size was 12 m × 12 m and only the central 5 m × 5 m plot was used for experimental purposes. Mean amounts of N, P and K over the crop rotation annually applied to the investigated treatments, basic soil chemical properties and concentrations of trace elements are given in Tables 1–3. Application of mineral fertilizers directly to spring barley was 70 kg N ( $N_4$  level), 60 kg P ( $P_2$  level) and 100 kg K ( $K_2$  level) per ha. Approximately 25% of N, P and K applied by organic fertilizers to sugar beet could be used by investigated spring barley in the following year. Amount of N, P and K applied by FM to sugar beet was 66, 15 and 159 kg/ha and amount of N, P and K applied by PL was 81, 117 and 354 kg/ha.

**Grain sampling, chemical and statistical analysis.** Half-late malting spring barley variety Sebastian was used in this study. The grain was gathered using a harvester and grain yield per ha was then determined. The samples of grain were collected from each plot during the harvest on 10<sup>th</sup> August 2010. The grain samples were washed three times with distilled water and then dried at 105°C up to total desiccation.

The concentrations of elements in the grain were determined by wet ashing with increased pressure. Exactly 1 g of powdered grain was decomposed in a mixture of 8 mL of  $HNO_3$  and 2 mL of  $H_2O_2$ . The ash was then decomposed using a microwave ash-

Table 1. List of investigated fertilizer treatments and mean annual application rates of N, P, K, Cu and Zn over all crops in nine years long crop rotation. Approximate amount of applied Cu and Zn by fertilizer treatments follows calculations by Upreti et al. (2009)

| Element   | Control | $N_4P_2K_2$ | $FMN_4P_2K_2$ | $PLN_4P_2K_2$ |
|-----------|---------|-------------|---------------|---------------|
| N (kg/ha) | 0       | 91          | 113           | 118           |
| P (kg/ha) | 0       | 31          | 36            | 70            |
| K (kg/ha) | 0       | 146         | 199           | 264           |
| Cu (g/ha) | 0       | 4           | 10            | 27            |
| Zn (g/ha) | 0       | 43          | 156           | 175           |

$N_4$  – ammonium nitrate;  $P_2$  – superphosphate;  $K_2$  – potassium chloride; FM – farmyard manure; PL – poultry litter

Table 2. Mean values of basic soil chemical properties in arable layer in 2007. Plant available P and K concentrations were determined by Mehlich III extraction procedure

| Chemical properties     | Control | N <sub>4</sub> P <sub>2</sub> K <sub>2</sub> | FMN <sub>4</sub> P <sub>2</sub> K <sub>2</sub> | PLN <sub>4</sub> P <sub>2</sub> K <sub>2</sub> |
|-------------------------|---------|--|--|--|
| pH (CaCl <sub>2</sub> ) | 6.0     | 6.3  | 5.8  | 5.4  |
| C organic (%)           | 2.6     | 1.41   | 3.9  | 3.4  |
| N total (%)             | 0.13    | 0.12   | 0.25   | 0.15   |
| C/N ratio               | 20      | 11.8   | 15.6   | 22.6   |
| P (mg/kg)               | 17      | 50   | 97   | 135  |
| K (mg/kg)               | 112     | 235  | 360  | 309  |
| Ca (mg/kg)              | 3100    | 3210   | 2840   | 2880   |
| Mg (mg/kg)              | 145     | 132  | 167  | 148  |

N<sub>4</sub> – ammonium nitrate; P<sub>2</sub> – superphosphate; K<sub>2</sub> – potassium chloride; FM – farmyard manure; PL – poultry litter

ing device (CEM 2000, Kamp, Lintfort, Germany) and diluted in 10 mL of *aqua regia*. As, Ca, Cd, Cr, Cu, K, Mg, Mn, Ni, P, Pb and Zn were measured by ICP-OES (Thermo Jarrell Ash, Trace Scan, Franklin, USA). To determine the N concentration, barley grain was mineralized in 98% H<sub>2</sub>SO<sub>4</sub> in microwave; N concentrations in the solution were then measured with a flow colorimeter (SAN plus SYSTEM, SKALAR, Breda, the Netherlands).

One way ANOVA followed by post-hoc comparison using Tukey test was used to identify significant differences among treatments.

## RESULTS AND DISCUSSION

There was a clear positive effect of fertilizer treatment on grain yield and concentrations of N, P and K in the grain, but no effect on concentrations of Ca and Mg (Table 4). With the exception of Fe, concentrations of micro- and risk-elements in the grain were not significantly affected by applied treatments (Table 4).

The yield of grain in the control (4.03 t/ha) was almost the same as the mean grain yield (4.07 t/ha) recorded in the Czech Republic in 2010 and higher

Table 3. Plant-available (extracted by CaCl<sub>2</sub>) and total (extracted by *aqua regia*) concentrations of trace elements (mg/kg) in arable layer (according to Upreti et al. 2009)

| Element | Extract           | Control | N <sub>4</sub> P <sub>2</sub> K <sub>2</sub> | FMN <sub>4</sub> P <sub>2</sub> K <sub>2</sub> | PLN <sub>4</sub> P <sub>2</sub> K <sub>2</sub> |
|---------|-------------------|---------|--|--|--|
| As      | CaCl <sub>2</sub> | 0.25    | 0.10   | 0.28   | 0.25   |
|         | total             | 13      | 10   | 12   | 12   |
| Cd      | CaCl <sub>2</sub> | < 0.01  | < 0.01                                       | 0.01   | 0.01   |
|         | total             | 0.69    | 0.70   | 0.72   | 0.71   |
| Cr      | CaCl <sub>2</sub> | < 0.01  | < 0.01                                       | < 0.01   | < 0.01   |
|         | total             | 59      | 59   | 59   | 59   |
| Cu      | CaCl <sub>2</sub> | 0.16    | 0.20   | 0.26   | 0.17   |
|         | total             | 14      | 14   | 15   | 15   |
| Mn      | CaCl <sub>2</sub> | 0.57    | 0.43   | 1.94   | 2.62   |
|         | total             | 486     | 482  | 491  | 518  |
| Ni      | CaCl <sub>2</sub> | 0.08    | 0.06   | 0.16   | 0.17   |
|         | total             | 21      | 21   | 21   | 22   |
| Pb      | CaCl <sub>2</sub> | 0.09    | 0.05   | 0.06   | 0.05   |
|         | total             | 24      | 25   | 25   | 27   |
| Zn      | CaCl <sub>2</sub> | 0.06    | 0.02   | 0.13   | 0.20   |
|         | total             | 53      | 54   | 55   | 62   |

N<sub>4</sub> – ammonium nitrate; P<sub>2</sub> – superphosphate; K<sub>2</sub> – potassium chloride; FM – farmyard manure; PL – poultry litter

Table 4. Concentrations of elements in grain of spring barley and yield of spring barley grain (85% dry matter content) in 2010

| Element                  | Control                  | N <sub>4</sub> P <sub>2</sub> K <sub>2</sub> | FMN <sub>4</sub> P <sub>2</sub> K <sub>2</sub> | PLN <sub>4</sub> P <sub>2</sub> K <sub>2</sub> |
|--------------------------|--------------------------|--|--|--|
| N (g/kg)*                | 15.3 <sup>a</sup> ± 1.12 | 20.0 <sup>b</sup> ± 1.37                     | 20.9 <sup>b</sup> ± 1.49                       | 21.6 <sup>b</sup> ± 1.21                       |
| P (g/kg)*                | 3.2 <sup>a</sup> ± 0.21  | 4.3 <sup>b</sup> ± 0.36                      | 4.5 <sup>b</sup> ± 0.19                        | 4.5 <sup>b</sup> ± 0.15                        |
| K (g/kg)*                | 4.4 <sup>a</sup> ± 0.32  | 5.0 <sup>b</sup> ± 0.23                      | 5.2 <sup>b</sup> ± 0.16                        | 5.0 <sup>b</sup> ± 0.22                        |
| Ca (g/kg) <sup>ns</sup>  | 0.6 ± 0.05               | 0.6 ± 0.07                                   | 0.6 ± 0.06                                     | 0.7 ± 0.05                                     |
| Mg (g/kg) <sup>ns</sup>  | 1.4 ± 0.07               | 1.5 ± 0.10                                   | 1.4 ± 0.08                                     | 1.4 ± 0.08                                     |
| As (mg/kg) <sup>ns</sup> | 0.07 ± 0.02              | 0.07 ± 0.02                                  | 0.04 ± 0.003                                   | 0.05 ± 0.02                                    |
| Cd (mg/kg) <sup>ns</sup> | 0.03 ± 0.004             | 0.01 ± 0.003                                 | 0.03 ± 0.005                                   | 0.01 ± 0.003                                   |
| Cr (mg/kg) <sup>ns</sup> | 0.40 ± 0.05              | 0.78 ± 0.22                                  | 0.19 ± 0.03                                    | 0.32 ± 0.11                                    |
| Cu (mg/kg) <sup>ns</sup> | 4.91 ± 0.35              | 6.01 ± 1.02                                  | 4.59 ± 0.16                                    | 4.23 ± 0.13                                    |
| Fe (mg/kg)*              | 49.8 <sup>a</sup> ± 7.85 | 75.6 <sup>b</sup> ± 3.97                     | 39.0 <sup>a</sup> ± 2.76                       | 36.0 <sup>a</sup> ± 3.23                       |
| Mn (mg/kg) <sup>ns</sup> | 16.3 ± 2.01              | 20.2 ± 3.49                                  | 13.2 ± 1.00                                    | 14.0 ± 0.51                                    |
| Ni (mg/kg) <sup>ns</sup> | 0.36 ± 0.06              | 0.42 ± 0.07                                  | 0.27 ± 0.07                                    | 0.23 ± 0.03                                    |
| Pb (mg/kg) <sup>ns</sup> | 0.14 ± 0.04              | 0.08 ± 0.02                                  | 0.18 ± 0.05                                    | 0.18 ± 0.02                                    |
| Zn (mg/kg) <sup>ns</sup> | 29.2 ± 2.68              | 30.8 ± 4.34                                  | 20.3 ± 0.46                                    | 26.7 ± 3.48                                    |
| Yield (t/ha)*            | 4.0 <sup>a</sup> ± 0.2   | 9.6 <sup>b</sup> ± 0.2                       | 9.7 <sup>b</sup> ± 0.2                         | 7.4 <sup>b</sup> ± 0.2                         |

<sup>ns</sup>Result of one-way ANOVA was not significant; \*result of ANOVA was significant at 0.05 probability level. Using Tukey post-hoc test, treatments with the same letter were not significantly different. ± value represents SE. N<sub>4</sub> – ammonium nitrate; P<sub>2</sub> – superphosphate; K<sub>2</sub> – potassium chloride; FM – farmyard manure; PL – poultry litter

than yields (1.9–3.3 t/ha) recorded over 12 years with no fertilizer input on different soil types in the Czech Republic (Černý et al. 2010). The relatively high yield in the control without any fertilizer input in 2010 was consistent with high yields in the controls of other field strips in the RFE in other years (5.0, 5.5 and 3.9 t/ha in 2008, 2009 and 2011, respectively), indicating the high suitability of the study site for barley production. The high suitability of the study site for barley production is also illustrated by grain yields over 9 t/ha recorded in the N<sub>4</sub>P<sub>2</sub>K<sub>2</sub> and FMN<sub>4</sub>P<sub>2</sub>K<sub>2</sub> treatments, which are very high in comparison with maximal grain yields from different experiments in the Czech Republic – 5.2 t/ha by Černý et al. (2010), 6.8 t/ha by Hřivná et al. (2009), 7.2 t/ha by Koutná et al. (2003) and 8.8 t/ha by Sedlář et al. (2011). The reduced grain yield in PLN<sub>4</sub>P<sub>2</sub>K<sub>2</sub> compared to the N<sub>4</sub>P<sub>2</sub>K<sub>2</sub> and FMN<sub>4</sub>P<sub>2</sub>K<sub>2</sub> treatments was probably due to adverse effects on barley growth from the high nutrient application in this treatment.

The clear positive effect of N application on N concentration in the barley grain is in accordance with Pettersson and Eckersten (2007). The opti-

mal concentration of N for malting (14–18 g/kg; Grashoff and D'Antuono 1997) was recorded only in the control. Application of 70 kg N/ha in the form of mineral fertilizer directly to barley was too high to achieve an acceptable compromise between the grain yield and suitable concentration of N in the grain for malting. We also recorded positive effects of P and K application on concentrations of P and K in the grain. Although P and K concentrations in barley grain have rarely been studied, they are believed to be highly important for the seeds for sowing as low concentrations of P and K in seeds in general can negatively affect the rapidity of germination and development of seedlings (Hejman et al. 2012a, Nadeem et al. 2012, White and Veneklaas 2012).

Although poultry litter supplied the soil with considerable amounts of Cu, Mn and Zn (Upřety et al. 2009), there was no effect of fertilizer treatment on concentrations of these elements in the grain. This can be explained by a dilution effect and by the low ability of barley to accumulate Cu (Belyaeva et al. 2005). Concentrations of Cu in our study were lower to or comparable with con-

centrations recorded by other authors in barley grain: 0.4–9.5 mg/kg by Belyaeva et al. (2005) 2.3–9 mg/kg by Smart et al. (1992) and 5.2–6.0 mg/kg by Cox et al. (2001). Concentrations of Mn recorded by other authors were the same or higher than the concentrations in our study: 11–69 mg/kg by Smart et al. (1992) and 12–13 mg/kg by Cox et al. (2001). Concentrations of Zn in our study were comparable with other authors: 29–58 mg/kg by Dudka et al. (1996), 19–41 mg/kg by Smart et al. (1992) and 24–26 mg/kg by Cox et al. (2001).

Significantly higher concentrations of Fe recorded in the  $N_4P_2K_2$  treatment than in the other treatments can be explained by the lowest organic matter content in the soil in this treatment, as the availability of metal is negatively related to the organic matter content in the soil (Singh et al. 2010). In our study, concentrations of Fe were substantially higher than concentrations of Fe, ranging from 7.4 to 9.1 mg/kg, in a study by Cox et al. (2001), or lower than the concentration of Fe 167 mg/kg recorded by Singh and Garg (2006). Different concentrations of Fe were probably caused by different soil pH, as the availability of Fe is high in acid soils (Hejerman et al. 2010).

Concentrations of Ni were similar to concentrations recorded in barley grain by Cox et al. (2001), which ranged from 0.26 to 0.54 mg/kg. Concentrations of As were considered to be in normal range as the same As concentrations were recorded in barley monitoring in UK (Cox et al. 2001). In all treatments, concentrations of Cd were about one order below the EU limit, which is 0.12 mg/kg. In our study, Cd concentrations were substantially lower than concentrations recorded by other authors on Cd contaminated soils: 0.12–0.7 mg/kg by Dudka et al. (1996), 0.24 mg/kg by Şekara et al. (2005) and 0.05–0.12 mg/kg by Kaniuczak et al. (2011). Low Cd concentrations in our experiment were because of the almost neutral soil reaction resulting in low Cd availability and low application rate of Cd by fertilizers (Upreti et al. 2009). We recorded low Cd concentrations in grain despite the total concentration of Cd in the soil exceeding the Czech limit. This indicates a lack of positive correlation between the total content of Cd in the soil and Cd concentration in the grain on neutral or slightly acid soils. No effect of fertilizer treatment on Cd concentration in the grain is in accordance with Singh and Myhr (1998).

A trend toward lower concentrations of Cr in treatments with application of organic fertilizers

was recorded despite total concentrations of Cr in the arable layer being the same in all treatments (Upreti et al. 2009). The increased availability of Fe due to the lower soil pH, together with the higher availability of P in the  $PLN_4P_2K_2$  and  $FMN_4P_2K_2$  treatments than in the control and  $N_4P_2K_2$  treatments, probably decreased Cr uptake, because Fe and P can act as Cr antagonists (Wallace et al. 1976).

Concentrations of Pb were below the EU limit which is 0.24 mg/kg, and were the same or lower than concentrations recorded by other authors: 0.4–2.2 mg/kg by Dudka et al. (1996) in a Pb contaminated area, 0.02–0.48 mg/kg by Zhao et al. (2004) during monitoring of grain quality in the UK and 1.84 mg/kg by Şekara et al. (2005) in a remediation experiment. Concentrations of Pb in our experiment approached the EU limit, although the total Pb concentrations in the soil and in the fertilizers used were far below the legislative limits (Upreti et al. 2009). Similarly, as in the case of high concentrations of Pb recorded in potato tubers in the study area in 2009 (Šrek et al. 2010), a possible explanation seems to be atmospheric deposition, since the study area is in close proximity to the airport. Although there was no significant effect of treatment on concentrations of Cr, Ni and Pb, mean concentrations between particular treatments were relatively different. The possible explanation of no significant results is high variability in concentrations within each treatment.

Finally we concluded that fertilizer application greatly increased grain yield and N, P and K concentrations in the grain, but there was almost no effect on the concentrations of Ca and Mg, micro- and risk-elements. The long-term use of organic and mineral fertilizers at normal application rates on mineral rich soils does not appear to represent any risk for contamination of cereals by risk-elements.

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