

## Mineralogy and Chemical Composition of Cryosols and Andosols in Antarctica

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### Abstract

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Variations in mineralogical and elemental composition of Turbic Cryosol (King George Island and James Ross Island), Skeletic Cryosol (J. Ross Island, the Trinity Peninsula), Leptic Andosols (Deception Island), and Skeletic Andosols (Deception Island) were studied. Significant differences in mineralogical composition in size fraction, vertical and horizontal position were determined by X-ray diffraction quantitative analysis. The differences were attributed to a variable degree of chemical weathering process. As a product of weathering were defined montmorillonite, chlorite, microcrystalline polymorphous silica, Fe oxides, and oxyhydroxides. Their content was increased towards the surface and in a finer fraction. Crystal thickness and size distribution of montmorillonite was measured by the Bertaut-Warren-Averbach technique and the results confirmed higher intensity of chemical weathering in Turbic Cryosols (J. Ross Island and King George Island). Feldspars and volcanic glass were the main phases found in soil samples from Deception Island. They were determined as phases of parent rock and indicated low intensity of chemical weathering. Clinoptilolite was identified in soil samples from J. Ross Island and Deception Island and its hydrothermal origin was supposed. Soil (sample AP) from Base General Bernardo O'Higgins Riquelme had significantly higher content of phosphate minerals, which was a result of ornithogenic activities. Macro elements content was consistent with mineralogical analyses. Turbic Cryosols were rich in iron, aluminium, magnesium, and manganese (> 2%). Skeletic Cryosols contained also iron, aluminium, magnesium, manganese, calcium, and sulphur (> 1%). Leptic Andosols were rich in iron and aluminium (> 2%). Soil properties and humic substances quality were evaluated. Results of chemical and physical analyses indicated that poorly developed soils of Antarctica varied in acidity, conductivity, humus content, and texture. Humic substances content and quality were low. Mainly mechanical (physical) and chemical weathering processes played major role in the soil development accompanied by low intensity of biological activity.

**Keywords:** King George Island; Deception Island; James Ross Island; weathering; mineralogical composition; clay mineral

Climatic conditions have changed in the past half century and therefore a considerable attention is paid to the human impact on the Antarctic environment. The retreat and disintegration of ice shelves together with habitat changes are evidence of environmental changes, therefore important new investigations in ice-free areas of Antarctica are needed (UGOLINI & BOCKHEIM 2008). The soils of

Antarctica have formed since the last deglaciation on different lithological materials (e.g. andesitic and basaltic lavas, breccias, fluvio-glacial and moraine deposits, Cretaceous deposits etc.) being affected by a number of pedogenic processes as quoted UGOLINI (1964), NAVAS *et al.* (2008), and BOCKHEIM *et al.* (2015). Keys Soils Taxonomy (Soil Survey Staff 2006) divided permafrost affected soils (Gelisol)

into three suborders: Histels, Tuberls, and Orthels. Turbel suborder includes soils with ice-cemented permafrost within 70 cm of the surface. The main soil forming processes are cryoturbation, physical (mechanical) weathering, chemical weathering (oxidation and salinization), podsolization, rubification, cambic process, ornithogenic activities, and humus accumulation (BEYER *et al.* 1997, 2000; LEE *et al.* 2007; UGOLINI & BECKHEIM 2008; SIMAS *et al.* 2008; HAUS *et al.* 2016). Also volcanic activity (e.g. nearby Deception Island) is a factor directly affecting soil formation processes. Many authors came to the conclusion that the intensity of pedogenic processes was decreasing with the increasing latitude. The relationship between soil formation process, landscape, and the glacial and volcanic activities is not well understood and has involved many recent investigators. The attention is also focused on a number of soil forming factors and different mineral substrata, which directly affect pedogenesis under extreme climatic conditions. The studies of Antarctic ecosystem showed that in an ice-free environment there are a large number of species of non-flowering plants (lichen, mosses, algae, fungi) and animals. Only two species of flowering plants are known: the Antarctic hair grass (*Deschampsia antarctica*) and the Antarctic pearlwort (*Colobanthus quitensis*). These are the main source of soil organic carbon. Humification process and formation of humic acids in spite of the absence of lignin containing plants was confirmed by WILSON *et al.* (1986), BEYER *et al.* (1997, 2000), and GAJDOŠOVÁ *et al.* (2003). According to KAISER and GUGGENBERGER (2000) and KLEBER *et al.* (2005, 2007) the soil organic carbon concentration is closely related not only to texture but also to the content of extractable Fe and Al and nitrogen compounds. Significant correlation was found between chemical composition, texture, mineral specific surface area, and organic carbon concentration (WANG & XING 2005; GRANDY & NEFF 2008; PIETRAMELLARA *et al.* 2009). Chemical composition of mineral-associated organic carbon seems to influence the mineralization rate of soil organic carbon. Higher mineralization rate is observed for labile and more aliphatic carbon fraction and lower rate is characteristic for aromatic and more resistant organic carbon fractions (MIKUTTA *et al.* 2006; SCHNEIDER *et al.* 2010).

The aim of the present study was to determine the degree of weathering of poorly developed Turbic or Skeletic Cryosols and Leptic or Skeletic Andosols formed

in the ice-free area of Antarctica by analyzing the elemental and mineralogical composition. Furthermore, basic soil physical and chemical properties, humic substances quality and content, crystal thickness, and size distribution of montmorillonite were evaluated.

## MATERIAL AND METHODS

**Regional settings.** STRELIN and SONE (1998) recorded the mean annual air temperature (in deglaciated area and at sea level) of  $-6.5^{\circ}\text{C}$  and the mean annual precipitation around 200 mm water equivalent at James Ross Island. Deception Island is an active volcano with polar maritime climate, the mean annual air temperature measured was  $-3^{\circ}\text{C}$  (ranging from  $+11^{\circ}\text{C}$  to  $-28^{\circ}\text{C}$ ) and the mean annual precipitation was 500 mm water equivalent. Extreme hot microclimates exist around steaming fumaroles and geothermally heated water (INACH 2005). King George Island had the mean annual surface temperature (at Bellingshausen station)  $-2.8^{\circ}\text{C}$  (KEJNA 1999) and the mean annual precipitation was 700 mm water equivalent (KEJNA *et al.* 2013).

One of the soil forming factors is the parent material – the rock (e.g. JENNY 1944). King George Island is formed from Jurassic to Pleistocene stratiform volcano-sedimentary complexes cut by numerous intrusions (Table 1) (BIRKENMAJER 1980; BIRKENMAJER *et al.* 1990). Deception Island is a large, active, mainly basalt-andesite Antarctic volcano at the western end of the Bransfield Strait, a very young ( $< 4$  Ma) marginal basin situated at the northern end of the Antarctic Peninsula (BAKER *et al.* 1975; SMELLIE 2001). James Ross Island is a large basaltic volcano. The James Ross Island outcrops are dominated by spectacular voluminous (tens of  $\text{km}^3$ ) far-travelled lava-fed deltas that mainly erupted in a glacial (i.e. ice sheet or ice cap) setting, although at least one delta and several tuff cones were marine-emplaced (SMELLIE *et al.* 2009). The lava, breccias, and glacial sediments overly poorly exposed Cretaceous fine clastic sediments (CRAME *et al.* 2004; JOHNSON & SMELLIE 2007). Because no eruptions have been observed, James Ross Island is generally regarded as inactive. However, the volcanic field is probably still active, as suggested by the very long periods of dormancy (hundreds of ka) and the presence of two essentially pristine scoria cones in the east of James Ross Island (SMELLIE *et al.* 2009).

**King George Island.** One soil sample (KGI) derived from poorly developed soil profile to 10 cm.

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**James Ross Island.** Three soil profiles (JRI 1, 2, 5a, 5b, 5c and 6a, 6b, 6c) with different soil horizons; the first soil profile with samples JRI 1a: depth 0–10 cm; JRI 1b: depth 10–20 cm; the second soil profile JRI 5a: depth 5–15 cm (massive stone pavement on surface); JRI 5b: depth 25–30 cm, and JRI 5c: depth 38–45 cm; the third soil profile JRI 6 (Table 2) with subsamples JRI 6a: depth 0–3 cm; JRI 6b: depth 12–20 cm; JRI 6c: depth 51–57 cm and three other soil samples (JRI 2–4) from shallow, undifferentiated soils to the depth of 10 cm. Soil profiles were limited by ice (permafrost) or by soil skeleton.

**Deception Island.** Three soil samples (samples DI 1, 2, and 4) from shallow soils limited by soil skeleton or rock (Table 2).

**The Trinity Peninsula.** One soil sample (sample AP from Chilean Base General Bernardo O'Higgins Riquelme) from shallow soil (near a colony of penguins) limited by rock.

Soils were classified according to the IUSS Working Group WRB (IUSS 2014). Horizon designation was done by JAHN *et al.* (2006). Sampling was performed during the Antarctica expedition 2011–2012. Soil samples were collected into plastic bags, air-dried at the Johann Gregor Mendel Base (James Ross Island), and stored at room temperature  $\leq 10^{\circ}\text{C}$  (VLČEK 2016). Characteriza-

tion of sampling localities, GPS coordinates, and basic soil properties are given in Tables 1 and 2 and Figure 1, examples of soil pit JRI6 and DI4 in Figure 2.

**Soil analysis.** Basic soil properties were determined by the commonly used standard methods. Soil reaction was determined by the potentiometric method in distilled water and in 1M KCl solution (1 : 2.5). Particle size analysis was determined by the pipette method. Total organic carbon content (TOC) was determined by wet digestion according to NELSON and SOMMERS (1982). Humic substances content was determined by the short fractionation method given by KONONOVA and BELTCHIKOVA (1963).

The X-ray diffraction (XRD) analysis: a corundum-spike sample was milled in methanol using a McCrone Micronising Mill (Retsch, Germany), then dried and side loaded to a XRD holder (EBERL *et al.* 1996; EBERL 2003; ŠRODOŇ *et al.* 2001). The XRD analysis was carried out in the  $4\text{--}65^{\circ} 2\Theta$  range using a Philips X-ray diffractometer model 1710 (Almelo, the Netherlands) equipped with Cu tube and Ni filter at 20 mA and 40 kV. The exposure was 2 s per step ( $0.02^{\circ} 2\Theta$ ). Quantitative analysis of 11 samples of soil fraction ( $< 2\text{ mm}$ ) and five samples of soil skeleton ( $> 2\text{ mm}$ ) was performed using the RockJock 11 program (EBERL 2003). To determine clay minerals, six soil samples

Table 1. Basic characteristics of the studied localities (see number of sample in Figure 1)

No.	Sample	Altitude (m)	Land cover	Parent material	Latitude	Longitude
1–2	James Ross Island (JRI 1a + 1b)	35	–	Cretaceous deposits undiff.	$63^{\circ}48.014'S$	$57^{\circ}53.662'W$
3	James Ross Island (JRI 2)	375	–	hyaloclastite tuffs	$63^{\circ}48.500'S$	$57^{\circ}56.923'W$
4	James Ross Island (JRI 3)	320	–	hyaloclastite breccias	$63^{\circ}49.270'S$	$57^{\circ}56.000'W$
5	James Ross Island (JRI 4)	55	–	Cretaceous deposits undiff.	$63^{\circ}48.335'S$	$57^{\circ}55.082'W$
6–8	James Ross Island (JRI 5a+5b+5c)	53	–	Cretaceous deposits undiff.	$63^{\circ}48.339'S$	$57^{\circ}55.082'W$
9–11	James Ross Island (JRI 6a+6b+6c)	38	–	glacial/fluvioglacial deposits	$63^{\circ}53.135'S$	$57^{\circ}57.692'W$
12	James Ross Island (JRI 7)	44	–	glacial/fluvioglacial deposits	$63^{\circ}53.405'S$	$57^{\circ}57.661'W$
13	Deception Island (DI 1)	20	–	post-caldera series, tuffs	$62^{\circ}58.748'S$	$60^{\circ}33.280'W$
14	Deception Island (DI 2)	12	–	post-caldera series, tuffs	$62^{\circ}58.715'S$	$60^{\circ}33.438'W$
15	Deception Island (DI 4)	4	–	post-caldera series, tuffs	$62^{\circ}58.742'S$	$60^{\circ}33.413'W$
16	King George Island (KGI)	50	–	Tertiary andesitic and basaltic lavas	$62^{\circ}12.187'S$	$58^{\circ}57.593'W$
17	Antarctic /Trinity Peninsula (AP)	10	–	metasediments	$63^{\circ}19.282'S$	$57^{\circ}53.893'W$

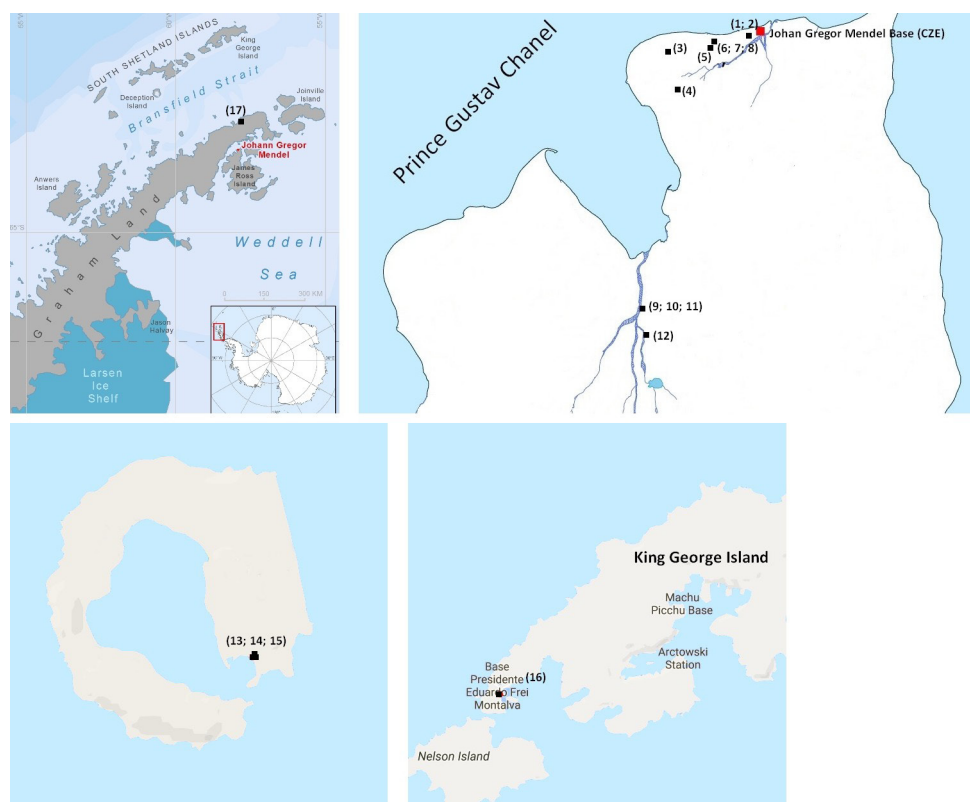


Figure 1. Geographical location of soil sampling; top row, from left to right: detail of Graham Land with islands (source: Zdeněk Stachoň, Masaryk University), detail of James Ross Island (source Czech Geological Survey 2009); bottom row, from left to right: Deception Island and King George Island (source: Google maps) (for more information see Table 1)

(< 2 mm) were selected, ground, and sieved to obtain fraction < 0.16 mm. Subsequently, the Jackson treatment modified by ŠUCHA *et al.* (1991) was applied to remove carbonates, organic matter, and Fe oxides and hydroxides. Clay fraction (< 2  $\mu\text{m}$ ) was separated by sedimentation in a water suspension following Stokes' law. Excess soluble salts were removed by dialysis. Orientated powder samples were prepared by the glass slide method (MOORE & REYNOLDS 1997) with drying of the suspensions at room temperature. Samples were analyzed in an air-dried state and saturated with ethylene glycol overnight at 70°C (EG) using the Philips PW 1710 diffractometer (Philips, Netherland) ( $\text{CuK}\alpha$  radiation, 35 kV, 20 mA, step size 0.2°, 1 s per step; measured interval was 2–50° 2 $\theta$ ). The Bertaut-Warren-Averbach (BWA) technique and MudMaster program for calculating mean crystallite thickness ( $T_{\text{MEAN}}$ ) and thickness distribution from the shapes of X-ray diffraction peaks according to EBERL *et al.* (1996), DRITS *et al.* (1998) and EBERL (2003) were applied. The interference functions were extracted from patterns by dividing the intensity by the Lorentz-polarization and layer structure factors

(DRITS *et al.* 1998). For the bulk chemical analysis an aliquot of each sample was dried overnight at 105°C, ground in a motorized agate mortar to a grain size of < 50  $\mu\text{m}$ , weighed, and digested in a microwave



Figure 2. Soil profile JRI 6 (samples JRI 6a, b, c) on James Ross Island and on Deception Island (sample DI 4) (from left to right)



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in an *aqua regia* solution. The solution was filtrated and analyzed by flame AAS (AAS Vario 6, Analytic Jena, Germany) for Fe. Macro elements content was determined by the Inductively Coupled Plasma Mass Spectrometry (ICP-MS) analysis at Bureau Veritas Commodities Canada Ltd.

## RESULTS AND DISCUSSION

### Pedological and geochemical characterization.

The studied soil profiles in Antarctica ice-free areas were poorly developed (57 cm maximum depth). Soils were mainly sandy loam (9 samples: JRI 1a, JRI 2, JRI 3, JRI 4, JRI 5c, JRI 6a, JRI 7, DI 1, KGI); loamy sand (3 samples: JRI 6c, DI 2, DI 4); loam (2 samples: JRI 1b and JRI 5a); and silt loam (2 samples: JRI 5b and JRI 6b) textured (Table 2). In the soil profiles JRI 5 and 6 the clay fraction has a maximum at a depth close to 10 cm; silt particles have a maximum at a depth 15–30 cm, and sand particles have two maxima: at the first subsurface horizon (because of stone pavement and selective wind and water erosion of fine fractions – OH & RICHTER 2005; BAUER & VELDE 2014) and deeper than 35 cm (because of weathering decreasing).

The studied soil samples varied in active soil reaction (3.6–7.9) and exchangeable soil reaction

(3.1–6.7). More acid were Turbic Cryosols JRI 5 and JRI 6 from James Ross Island mainly at deeper layers (Table 2, samples JRI 5c, JRI 6b, and JRI 6c). The electric conductivity (EC) of part of the studied samples showed similar values (100–200  $\mu\text{S}/\text{cm}$ ; Table 2) as published by NAVAS *et al.* (2008) and HAUS *et al.* (2016). Some samples from James Ross Island, however, presented 10-times higher values (Table 2). High EC (in samples with very low pH) can result by different ionic species in the soil solution. Ion  $\text{H}^+$  is the ion with the highest equivalent ionic conductance having (to low pH) the highest contribution to EC. We expected that low pH and high EC were caused precisely due to the high concentration of  $\text{H}^+$  ions but the absence of water soluble salts. EC was very low in Andosols (samples DI) as well as in the Turbic Cryosol formed from Tertiary andesitic and basaltic lavas (sample KGI, Table 2).

Total organic carbon (TOC) content in all samples was very low except the sample from the Trinity Peninsula (sample AP; Figure 3). The lowest values of TOC were in the samples from Deception Island (0.04%), TOC ranged from 0.09 to 0.92% at James Ross Island, decreased with depth (JRI 1), or was relatively stable in two other profiles (JRI 5, JRI 6; Table 2) (POSPÍŠILOVÁ *et al.* 2017). The highest value of TOC (4.94%) was in the sample from the Trinity

Table 2. Basic soil properties in fraction < 2 mm; values referring to samples 1–12 (except pH/KCl) were published by VLČEK (2016)

No.	Soil types (samples)	Depth (cm)	Clay <sup>1</sup>	Silt <sup>2</sup>	Sand <sup>3</sup>	TOC	pH/H <sub>2</sub> O	pH/KCl	EC ( $\mu\text{S}/\text{cm}$ )
			(%)						
1	Turbic Cryosol (JRI 1a)	0–10	7.8	21.0	71.2	0.70	7.9	6.3	116
2	Turbic Cryosol (JRI 1b)	10–20	14.8	36.2	49.0	0.29	7.9	6.5	133
3	Skeletal Cryosol (JRI 2)	0–10	1.4	32.9	65.6	0.21	7.9	6.7	523
4	Turbic Cryosol (JRI 3)	0–10	13.0	28.7	58.3	0.36	7.9	6.5	433
5	Turbic Cryosol (JRI 4)	0–10	6.7	39.2	54.1	0.92	6.4	5.2	815
6	Turbic Cryosol (JRI 5a)	5–15	18.8	35.0	46.2	0.08	7.9	6.6	200
7	Turbic Cryosol (JRI 5b)	25–30	2.1	52.0	45.9	0.13	7.2	6.6	1040
8	Turbic Cryosol (JRI 5c)	38–45	9.8	18.1	72.1	0.09	3.8	3.1	1480
9	Turbic Cryosol (JRI 6a)	0–3	15.3	21.5	63.2	0.25	7.0	6.5	2970
10	Turbic Cryosol (JRI 6b)	12–20	3.3	60.4	36.4	0.23	4.1	3.3	2852
11	Turbic Cryosol (JRI 6c)	51–57	4.3	22.4	73.3	0.38	3.6	3.1	1633
12	Turbic Cryosol (JRI 7)	0–10	14.1	25.0	60.9	0.30	7.0	6.4	3129
13	Leptic Andosol (DI 1)	0–10	2.4	22.4	75.2	0.04	7.2	5.1	n/a
14	Leptic Andosol (DI 2)	0–10	3.9	15.3	80.8	0.04	6.9	4.9	n/a
15	Skeletal Andosol (DI 4)	0–10	3.4	12.4	84.2	0.35	5.8	3.9	n/a
16	Turbic Cryosol (KGI)	0–10	14.9	25.1	60.0	0.54	7.2	5.3	n/a
17	Skeletal Cryosol (AP)	0–10	n/a	n/a	n/a	4.94	5.4	4.9	277

<sup>1</sup>clay: < 0.002 mm; <sup>2</sup>silt: 0.05–0.002 mm; <sup>3</sup>sand: 2.00–0.05 mm; n/a = not analyzed; TOC – total organic carbon; EC – electrical conductivity

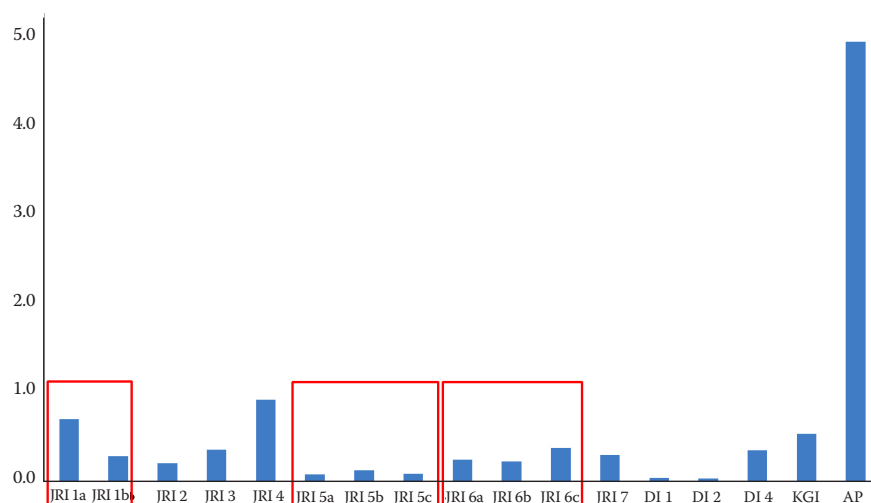


Figure 3. Total organic carbon (TOC, %) content in studied soils (for description see Table 1)  
Soil profiles are labelled by red frame

Peninsula because of nesting activities of penguins. Similar TOC contents were published by NAVAS *et al.* (2008) and HAUS *et al.* (2016) in the soil samples from the Hurd Peninsula, Livingston Island. Organic material was derived from different sources and was spatially variable but generally the humification degree was between 20–30%, which can be evaluated according to ORLOV (1985) as a slow humification process. Similar data about low humus content and quality were published by BOCKHEIM and UGOLINI (1990), BOCKHEIM (1997) and BOCKHEIM *et al.* (2015). Also BEYER *et al.* (2000), SIMAS *et al.* (2008), and HAUS *et al.* (2016) showed that low humus content

is typical for young and poorly developed Antarctic soils. They stressed that humus content is highly spatial variable according to the mineral substrata quality, latitude, altitude, and faunal activities. Data on the humus content in the studied soil profiles indicated young and poorly developed soils at all studied localities. Especially Leptic Andosols (Deception Island) formed on andesitic and basaltic lavas.

**Deception Island** (samples DI): the samples contained 2.0–2.3% of iron, 1.3–1.6% of aluminium, 0.9% of magnesium, and 0.8–1.4% of calcium. The contents of potassium were 0.08–0.1%, sodium 0.5–0.6%, phosphorus 0.03–0.1%, and sulphur less than 0.05% (Figure 4).

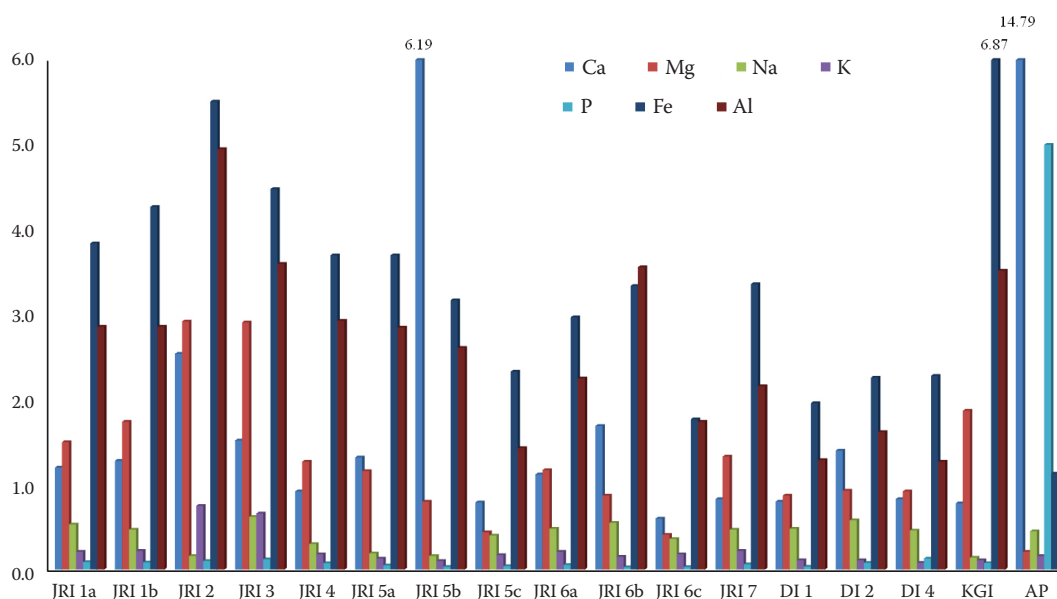


Figure 4. Macro elements content (%) in the studied soil types (for description see Table 1)

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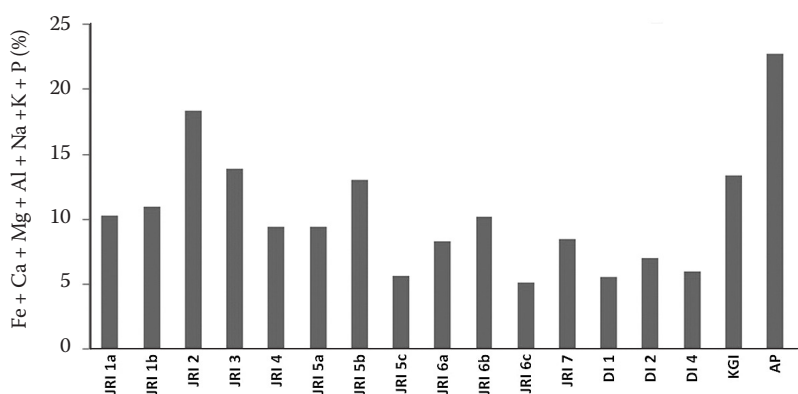


Figure 5. The sum of macro elements of studied soil samples determined after treatment in the *aqua regia* solution

**King George Island** (sample KGI): the sample was rich in iron (6.9%) and aluminium (3.5%). The content of magnesium was lower than 1.9%, the content of calcium 0.9%, of potassium 0.1%, of sodium 0.1% (Figure 4).

**The Trinity Peninsula** (sample AP): the sample contained 1.1% of iron, 1.0% of aluminium, 0.2% of magnesium but 14.8% of calcium and more than 5% of phosphorus. The high content of calcium and phosphorus is a result of penguins nesting activity in the vicinity and the presence of phosphate minerals (apatite and other phosphate minerals – see below). The content of potassium was 0.2% and of sodium 0.5% (Figure 4). The sample AP had the highest sum of determined macro elements (Ca + Mg + Na + K + Fe + Al) of the samples analyzed (Figure 5) due to the presence of soluble phosphate minerals.

**James Ross Island** (samples JRI): the samples contained less iron (1.8–5.5%) than those from King George Island; the content of iron decreased with depth. The contents of aluminium were 1.4–5.0%, of magnesium 0.4–2.9%, of calcium 0.6–2.5% (except the sample JRI 5b with 6.2% of calcium at a 25–30 cm depth). The contents of potassium were 0.1–0.8%, of sodium 0.3–1.6%, of phosphorus 0.03–0.1%.

The sum of determined macro elements of the samples from James Ross Island and King George Island was in the range 8.25–18.37%, except two samples (JRI 5c and JRI 6c) from the bottom of soil profiles JRI 5 and JRI 6 (Figure 5). The values were 5.09 and 5.6%. The sums of determined macro elements indicate reducing weathering intensity downward through profiles. They also demonstrate more intensive weathering in James Ross Island and King George Island in comparison with soil samples from Deception Island (5.54–6.98%; Figure 5).

The increase of Al, Fe, Mg towards the top of soil profile at JRI 5 and JRI 6 can be related to chemical

weathering (Figure 4). The distribution of Ca and Na down the profile is not uniform. It seems that the content of these elements also decreases in bottom layers in comparison with the middle part of the soil profiles (Figure 4).

**Mineralogical characterization.** Mineralogical composition of the studied soils was summarized in Table 3. First, mineralogy of upper soil horizons (0–15 cm) from different sites and soil types was compared. The composition of major mineral phases of soils from J. Ross and King George Islands (Turbic and Skeletic Cryosols) significantly differed from those derived from Deception Island (Skeletic and Leptic Andosol) and Base General Bernardo O'Higgins Riquelme (sample AP – Skeletic Cryosol). Sample AP contained apatite or other phosphate minerals as dominant phase, probably a product of penguin guano (MYRCHA & TATUR 1991; HARBEN & KUŽVART 1996; SIMAS *et al.* 2007). Other significant minerals were Na-plagioclase, dioctahedral mica, Fe-oxyhydroxides and amorphous phase (all slightly less than 10 wt.%). Deception Island soil was rich in Ca-plagioclase and amorphous phases. On the other hand, major phases from J. Ross Island were montmorillonite, quartz, feldspars, and clinoptilolite. Though, quantitative composition of these major phases is variable depending on locality (Table 3). The King George Island soil composition was similar with the J. Ross Island soils but clinoptilolite was absent and instead of it higher amount of chlorite was encountered.

The mineral compositions of two fractions (< and > 2 mm) of top soil profiles were compared to distinguish the degree of chemical weathering. Assumptions were that the amount of original minerals from parent rocks in the soil skeleton is decreasing with increasing weathering intensity and the products of chemical weathering are accumulating in fine fraction. In other words, we supposed high differ-

ences in mineralogy of the two studied fractions if the chemical weathering would take place in a more significant way. The differences pointing to chemical weathering were detected in most phases. The soil skeleton had higher contents of feldspars and volcanic glass. The fraction < 2 mm had higher contents of montmorillonite, sum of all phyllosilicates and clinoptilolite; all these phases, in general, are alteration products of feldspars and volcanic glass. The mineral transformations are a result of chemical weathering or hydrothermal alterations (CAMPBELL & CLARIDGE 1987). Exceptions are also present, the similar or opposite mineral distribution, they are related to quartz and partly to clinoptilolite (Table 4). The amount of quartz is practically the same in both studied fractions except for sample JRI 6a. Quartz is resistant to chemical weathering and its accumulation in the finer fraction is possible due to intensive physical weathering and psammitic char-

acter of parent rock. The contents of clinoptilolite are the same in both fractions of the sample JRI 5a and slightly higher in the coarse fraction in JRI 1b (Table 4). It could be explained by two different processes: intensive chemical weathering has wiped off the difference between the fractions or clinoptilolite is not a product of weathering but of low-temperature hydrothermal alteration. Zeolite minerals, but other than clinoptilolite, predominately phillipsite and chabasite, are secondary minerals that have been precipitated during short-term hydrothermal episodes related to the formation of volcanoclastic deposits in several areas of Antarctica e.g. on James Ross Island (JOHNSON & SMELLIE 2007), on Minna Bluff, a 45 km long volcanic peninsula extending southeast into the Ross Sea (ANTIBUS *et al.* 2014).

The mineralogical analysis vertical soil profiles was the next step aimed at distinguishing the origin of clay minerals, mainly montmorillonite and zeolite

Table 3. Mineralogical composition of soil fraction (< 2 mm); for explanations see Tables 1 and 2

Samples	JRI 1a	JRI 1b	JRI 5a	JRI 5b	JRI 5c	JRI 6a	JRI 6b	JRI 6c	DI 4	KGI	AP
Depth (cm)	0–10	10–20	5–15	25–30	38–45	0–3	12–20	51–57	0–10	0–10	0–10
Quartz	10.8	9.4	6.9	5.5	24.3	22.1	6	30.6	–	12.3	4.6
K-feldspar	12.2	9.8	3.3	2.0	2.0	11.2	6.1	11.7	6.0	3.0	5.0
Ca-plagioclases	19.3	17.1	15.7	11.0	22.7	15.0	3.5	7.7	30.3	17.3	4.0
Na-plagioclases	7.0	7.0	1.8	0.6	6.5	7.2	0.9	11.3	1.3	12.9	8.4
Calcite	–	0.5	1.2	13.0	0.4	0.6	0.1	–	1.2	0.5	1.0
Gypsum	0.4	0.4	0.7	6.0	2.0	0.8	3.7	1.3	2.0		5.0
Goethite	–	–	1.0	0.3	–	–	0.3	–	–	0.2	–
Clinoptilolite	11.7	11.8	9.8	7.3	12.1	7.6	18.5	13.5	6.0	0.8	3.0
Opal CT	0.8	0.4	2.2	2.7	0.6	0.6	1.9	0.3	2.5	0.5	1.5
Tridymite	1.0	1.5	5.0	7.0			3.2		2.9	2.4	–
Volcanic glass	4.5	4.3	1.9	4.0	4.0	5.4	4.5	2.5	22.0	2.4	8.5
Ilmenite	–	0.1	0.1	0.2	0.1	0.2	–	0.1	1.0	0.2	4.6
Hematite	0.1	–	–	–	–	–	–	–	0.2	1.5	–
Olivine	0.6	–	1.0	0.7	–	0.9	0.2	–	1.7	0.2	–
Analcime	3.8	2.5	–	–	–	1.4	0.2	0.3	1.0	0.2	2.6
Apatite	–	–	–	–	–	–	–	–	–	–	24.5
Pyroxene	1.3	1.8	1.2	–	0.3	1.1	0.5	–	3.0	1.8	1.5
Amphibole	–	–	0.4	0.5	–	–	–	–	1.5	–	5.0
Ferrihydrite	2.1	2.9	3.0	0.4	–	2.1	–	–	5.0	3.0	8.0
Kaolinite	1.0	0.5	1.5	0.2	0.3	0.5	0.2	0.3	1.5	4.5	–
Montmorillonite	10.6	13.3	30.5	27.5	10.2	11.6	37.0	11.0	3.0	18	–
Muscovite (illite)	4.0	7.7	7.4	6.0	13.2	6.5	9.1	7.5	3.0	4.3	9.0
Biotite	2.1	1.9	0.2	0.3	–	1.0	0.3	–	1.0	0.5	–
Chlorite ± berthierine	5.0	7.1	3.3	4.0	0.4	3.8	3.5	1.7	3.3	13.0	3.0
Full pattern degree of fit	0.084	0.084	0.0701	0.077	0.102	0.080	0.072	0.081	0.100	0.082	0.222



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Table 4. Comparison of major mineral phases of soil fraction (< 2 mm) and skeleton (> 2 mm); for explanations see Tables 1 and 2

Samples/depth	JRI 1b (10–20 cm)		JRI 5b (0–10 cm)		JRI 6a (0–3 cm)		DI 4 (0–10 cm)		KGI (0–10 cm)	
	< 2	> 2	< 2	> 2	< 2	> 2	< 2	> 2	< 2	> 2
Fraction (mm)										
Quartz	9.4	8.9	6.9	6.0	22.1	9.1	–	–	12.3	13.2
Feldspars	33.9	41.4	20.8	21.4	33.4	37.8	37.6	44.5	33.2	40.6
Clinoptilolite	11.8	13.9	9.8	9.4	7.6	4.9	6.0	–	0.8	0.5
Modification of SiO <sub>2</sub> without quartz	1.9	2.8	7.2	6.8	0.6	0.8	5.4	1.9	2.9	2.7
Volcanic glass	4.3	2.3	1.9	5.1	5.4	9.0	22.0	35.2	2.4	2.4
Clay minerals and other phyllosilicates	30.5	19.7	42.9	42.6	23.4	17.2	11.8	0.8	40.3	33.8

mineral – clinoptilolite; three profiles were examined. Small differences were observed in profile JRI 1, probably the profile was too shallow (only 20 cm thick). The remaining profiles were deeper, 45 and 57 cm, respectively and showed significant variances (Table 3, Figures 6 and 7). The profile JRI 5 showed features of chemical weathering, the amount of potential weathering products (phyllosilicates – mainly montmorillonite and chlorite, microcrystalline polymorphous silica, Fe-oxyhydroxides) increasing towards the surface and initial parent phases (feldspars, volcanic glass, quartz, olivine) decreasing in the same direction. The higher amount of calcite and gypsum in the horizon 25–30 cm also pointed to chemical weathering. The amount of clinoptilolite was similar in the entire profile (Table 3, Figures 6 and 7). A slightly different

situation was found in profile JRI 6 (JRI 6a, b, c). A zig-zag trend of mineral distribution in the studied parts of the soil profile (Figure 7) could be due to washout of clay particles from the uppermost soil layer. A lower amount of clay fraction in this layer was observed by particle size distribution (Table 2). The rest of the profile, however, displayed the same trend as profile JRI 5 (Figures 6, 7).

The major mineral of the clay fraction in samples from J. Ross Island (Turbic Cryosols) is montmorillonite (Tables 3, 5). Montmorillonite is a type of dioctahedral smectite and it was confirmed by Fourier Transform Infrared Spectroscopy and a comparison with Source Clays database (MADEJOVÁ & KOMADEL 2001). Sample JRI 5a was composed of almost pure montmorillonite with only traces of quartz and illite

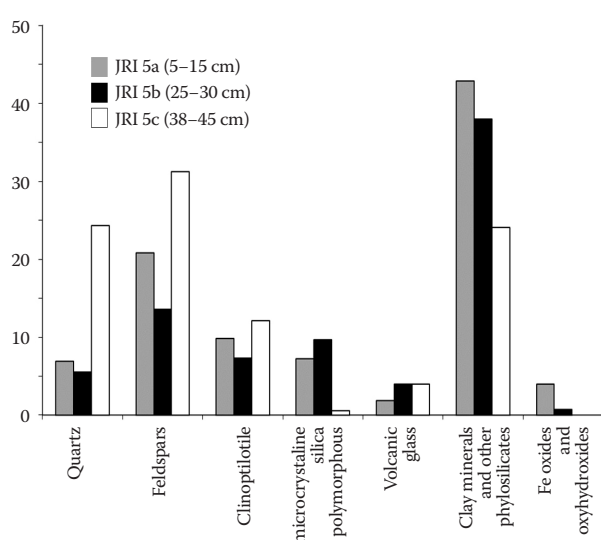


Figure 6. Major phases of mineralogical composition in soil profile JRI 5 (samples JRI 5a, b, c) of Turbic Cryosol (J. Ross Island) in fraction < 2 mm (in %)

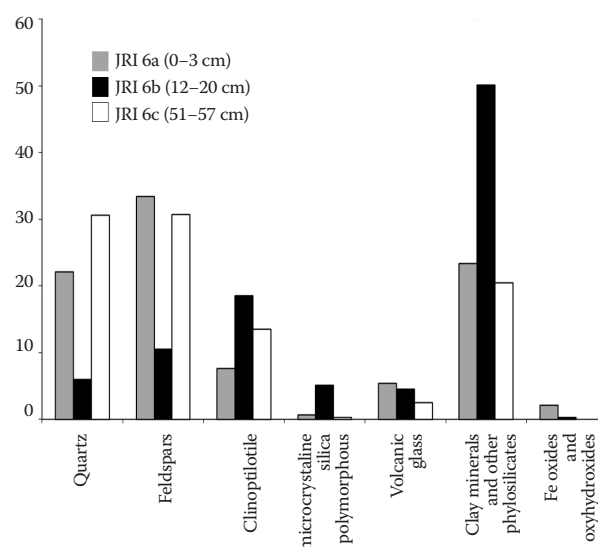


Figure 7. Major phases of mineralogical composition in soil profile JRI 6 (samples JRI 6a, b, c) of Turbic Cryosol (J. Ross Island) in fraction < 2 mm (in %)

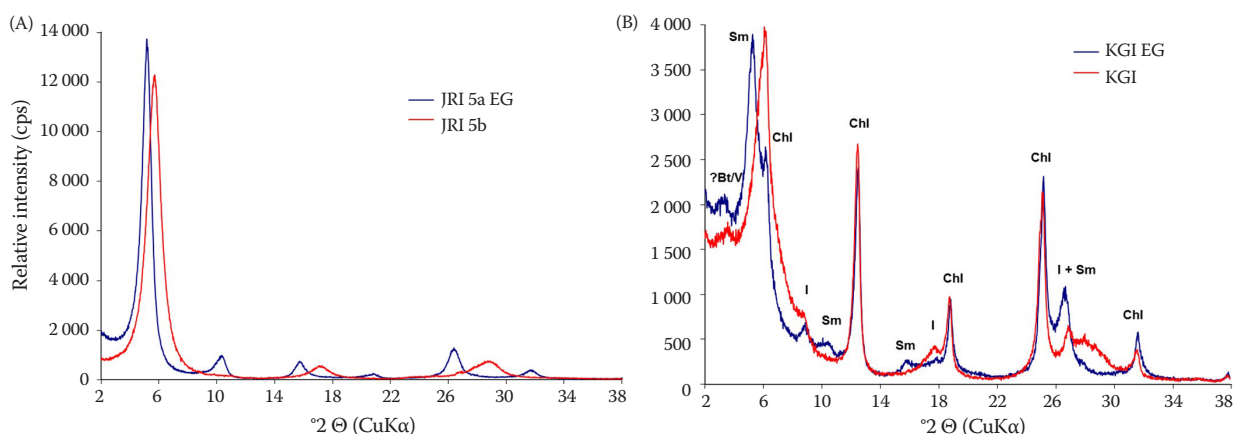


Figure 8. Selected X-ray diffraction patterns of oriented specimens of clay fraction: sample JRI 5a (A), sample KGI (B) in air-dried state (AD) and after ethylene glycol saturation (EG); Bt/V – interstratified biotite-vermiculite; Sm – smectite; I – illite; Chl – chlorite; ? – presence of the indicated mineral phase is questionable

(Figure 8A). The dominant mineral assemblage of the clay fraction, however, was polymineral. Along with montmorillonite, chlorite was found as the major mineral in part of the samples from J. Ross Island. The other minerals of the clay fractions were illite, clinoptilolite, quartz, and feldspars (Table 5). The sample from King George Island (Turbic Cryosols) had the same mineral assemblage except of an additional mineral phase, interstratified biotite-vermiculite (Table 5, Figure 8B).

The XRD patterns of the clay fraction with dominance of montmorillonite were submitted to BWA

analysis. Mean crystallite thickness ( $T_{\text{MEAN}}$ ) of all the nine montmorillonite samples analyzed was in the range of 5.56 to 7.19 nm and the distributions of montmorillonite thickness showed lognormal shape characteristic for smectite (Table 5, Figure 9).  $T_{\text{MEAN}}$  showed that the studied smectites belong to thinner, less developed crystals (MYSTKOWSKI *et al.* 2000, SIMIĆ & UHLÍK 2006). A relationship between  $T_{\text{MEAN}}$  and the condition of smectite crystals growth was also confirmed in our study, e.g. the least developed soil sample DI 4 from Deception Island (Leptic Andosol) had a low amount of montmorillonite and also the

Table 5. Clay fraction mineral composition of selected soil samples and mean thickness of smectite (montmorillonite) crystallite determined by the The Bertaut-Warren-Averbach (BWA) technique

Sample	Depth (cm)	Mineralogy of clay fraction	$T_{\text{MEAN}}$ (nm)
JRI 1a	0–10	smectite > illite, chlorite, clinoptilolite, quartz	6.18
JRI 1b	10–20	smectite > illite, chlorite, clinoptilolite, kaolinite? > quartz	6.60
JRI 3	0–10	smectite, chlorite, illite, kaolinite? > clinoptilolite, quartz	
JRI 4	0–10	smectite > illite, chlorite, kaolinite, clinoptilolite, > quartz, feldspar	5.56
JRI 5a	5–15	smectite >>> quartz, illite	7.04
JRI 5b	25–30	smectite > illite, chlorite, clinoptilolite > quartz	6.73
JRI 5c	38–45	smectite > illite, chlorite, clinoptilolite > quartz	6.37
JRI 6a	0–3	smectite, chlorite > illite, kaolinite?, clinoptilolite, quartz, feldspar	
JRI 6b	12–20	smectite >> illite, chlorite, clinoptilolite, quartz	7.19
JRI 6c	51–57	smectite >> illite, chlorite, clinoptilolite, quartz	6.21
JRI 7	0–10	smectite, chlorite > illite, kaolinite?, clinoptilolite, quartz	
DI 4	0–10	albite > smectite	5.56
KGI	0–10	chlorite ≥ smectite, illite > mixed-layered biotite/vermiculite? > quartz	
AP	0–10	apatite >> quartz, feldspars, illite, amphibole	

? – presence of the indicated mineral phase is questionable;  $T_{\text{MEAN}}$  – mean crystallite thickness

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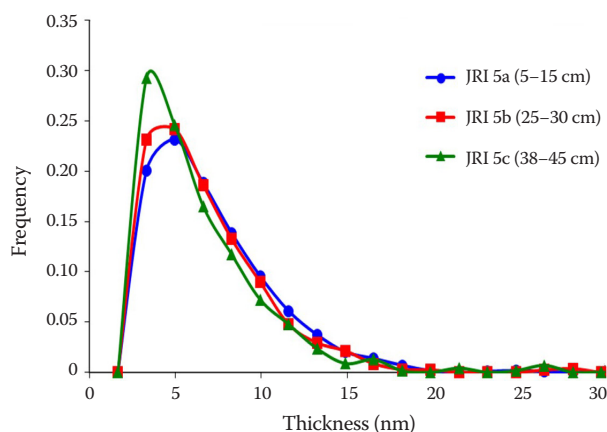


Figure 9. Log-normal thickness size distribution of montmorillonites from soil profile JRI 5 (samples JRI 5a, b, c) determined by the BWA technique

lowest  $T_{\text{MEAN}}$  (Tables 3 and 5). A slight increasing of montmorillonite  $T_{\text{MEAN}}$  towards the surface was detected in all the studied soil profiles (Table 5, Figure 9). It demonstrates the growth of montmorillonite crystals in the direction of soil evolution. This trend elucidates the origin of montmorillonite and leads to chemical weathering. An opposite development was observed in the soil profiles on bentonite deposits in temperate climates (ŠUCHA *et al.* 2001; STRÍČEK *et al.* 2006). In these cases, chemical weathering reduced  $T_{\text{MEAN}}$  of montmorillonites primarily originating by diagenetic and/or hydrothermal alteration of volcanic glass. The mentioned studies indirectly indicate that the growth of Antarctic montmorillonites was influenced predominantly by chemical weathering and not hydrothermal alteration. In general, the intensity of hydrothermal alteration decreased towards the surface.

## CONCLUSION

The detail mineralogical analysis of soil samples from the islands near the north-eastern extremity of the Antarctic Peninsula was applied to evaluate the presence of chemical along with physical weathering. We assumed the highest degree of chemical weathering to occur at surface horizons (higher temperature and presence of liquid water). The increased amount of potential weathering products, phyllosilicates (mainly montmorillonite and chlorite), microcrystalline polymorphous silica, Fe-oxyhydroxides, and the decreased number of initial parent phases (feldspars, volcanic glass, quartz, olivine) in fraction < 2 mm and in the upper soil horizons confirmed the process of chemical weathering. The dominant amount of montmorillonite

in the clay fraction was used to determine the crystal thickness using the BWA technique. Mean thickness of crystal in soil profiles rising towards the surface suggests the effect of chemical weathering. The isotopic evaluation of montmorillonite could provide further proof of chemical weathering in future research. The origin of zeolite mineral – clinoptilolite, in the studied samples present mostly at ca. 10 wt %, has remained undisclosed. Clinoptilolite could be a product of hydrothermal alteration, chemical weathering or both. The question of its origin is also a subject for future research.

The distribution of macro elements of the studied soil samples confirms more chemical weathered material in the soil profiles of James Ross Island. It also indicates more intensive or longer weathering on James Ross and King George Islands than on Deception Island.

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